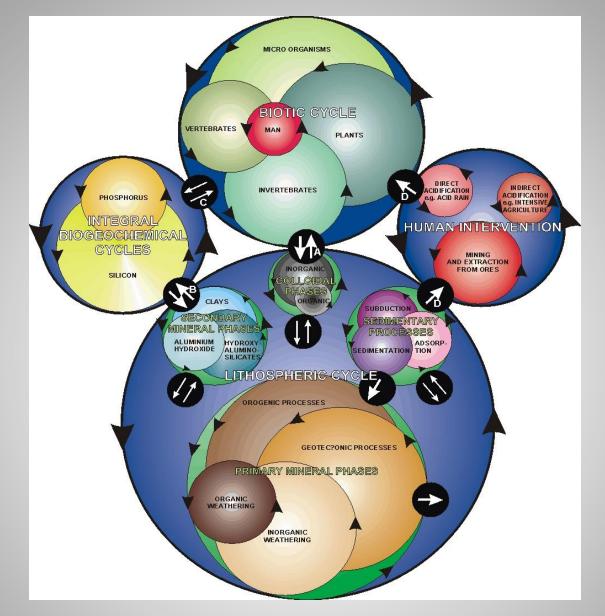
Human Exposure to Aluminium Aluminium Age and/or Aluminium Tyranny?

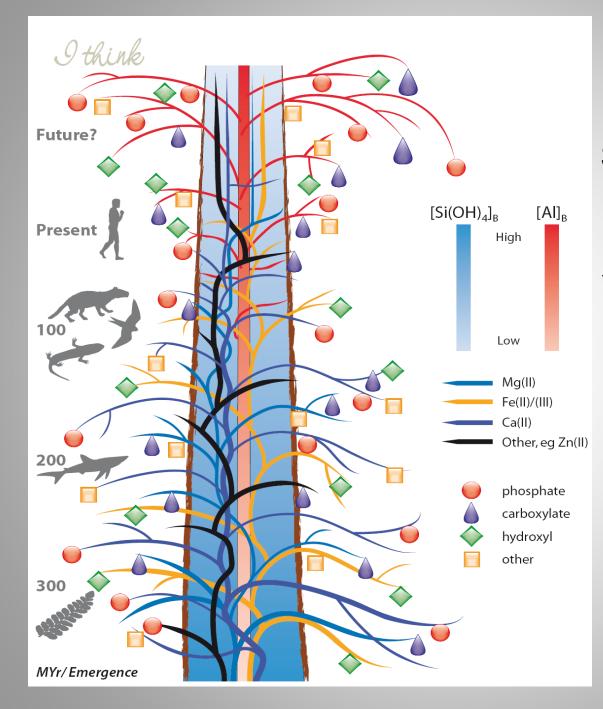
Christopher Exley PhD FRSB Professor of Bioinorganic Chemistry Aluminium and Silicon Research Group The Birchall Centre, Lennard-Jones Laboratories, Keele University, Staffordshire, ST5 5BG, UK <u>c.exley@keele.ac.uk</u> <u>http://www.keele.ac.uk/aluminium/</u>

Honorary Professor, University of the Highlands and Islands, Scotland, UK



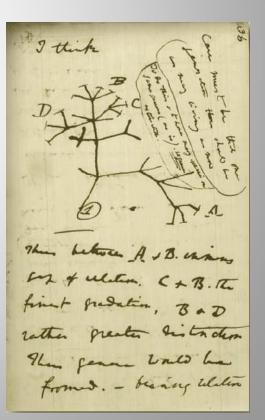
THE BIOGEOCHEMICAL CYCLE OF ALUMINIUM

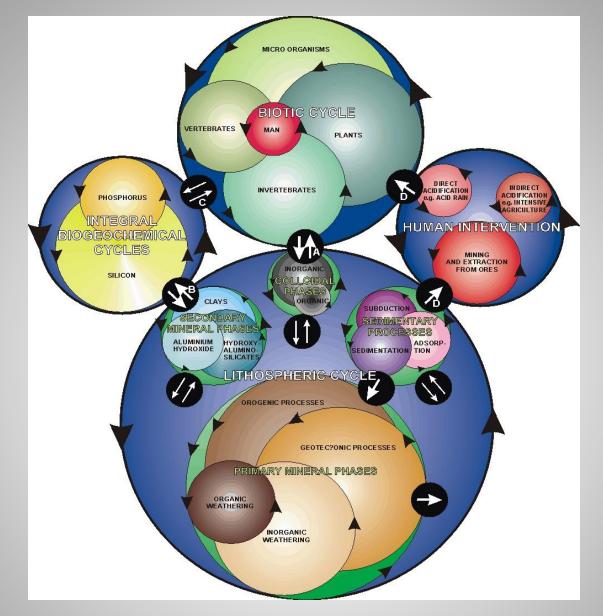
Exley C (2003) A biogeochemical cycle for aluminium ? J. Inorg. Biochem. 97, 1-7.



A Biochemical 'Tree of Life' for the Natural Selection of Aluminium

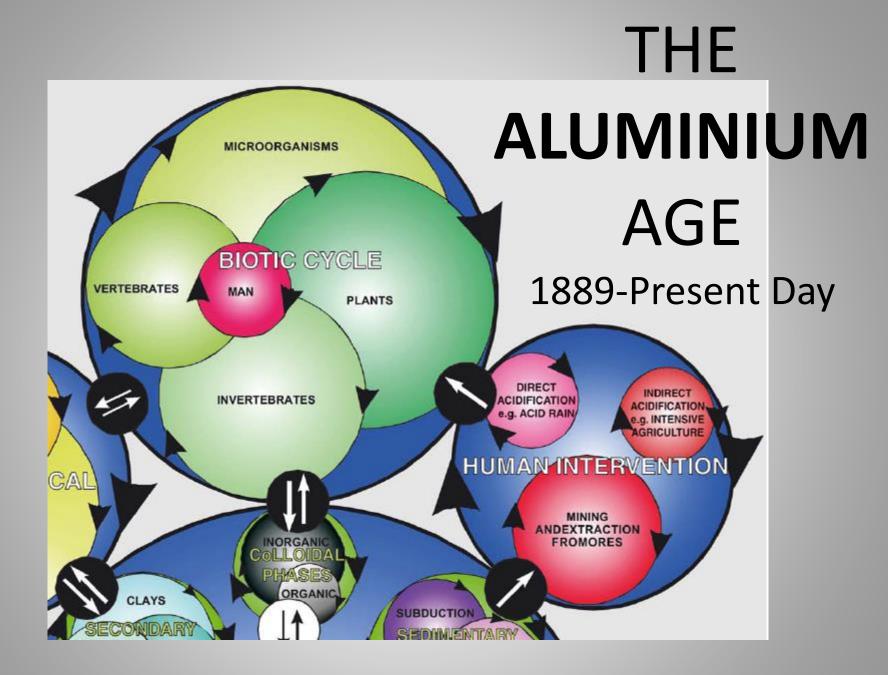
Exley C (2009) Darwin, natural selection and the biological essentiality of aluminium and silicon. Trends in Biochemical Sciences 34, 589-593.

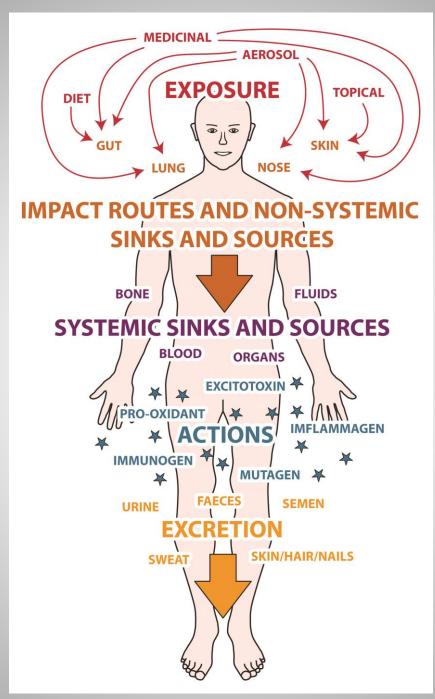




THE BIOGEOCHEMICAL CYCLE OF ALUMINIUM

Exley C (2003) A biogeochemical cycle for aluminium ? J. Inorg. Biochem. 97, 1-7.



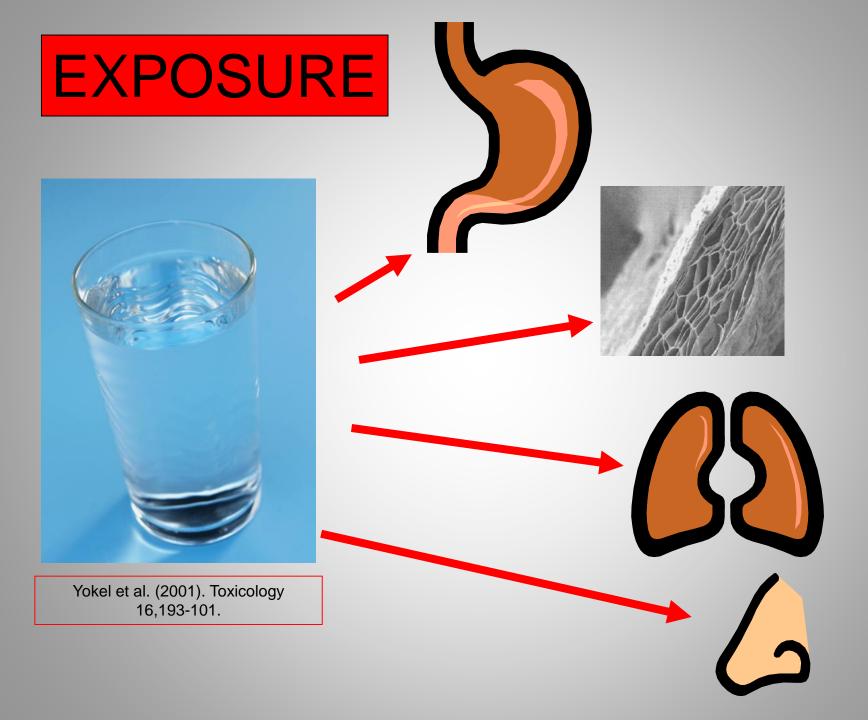


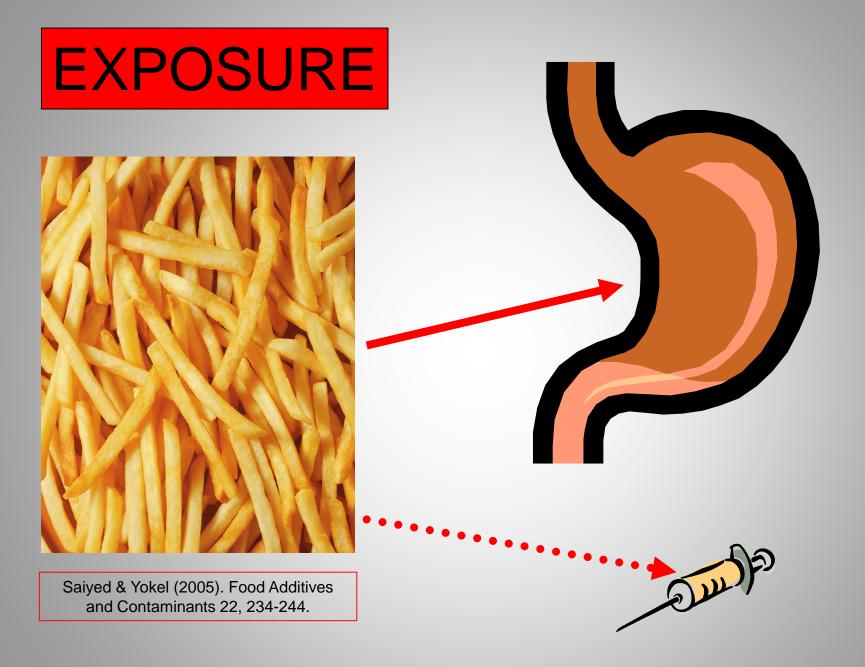
Exley C (2013) Human exposure to aluminium. Environmental Science:Processes and Impacts 15, 1807-1816.

The Body Burden of Aluminium: What is it?







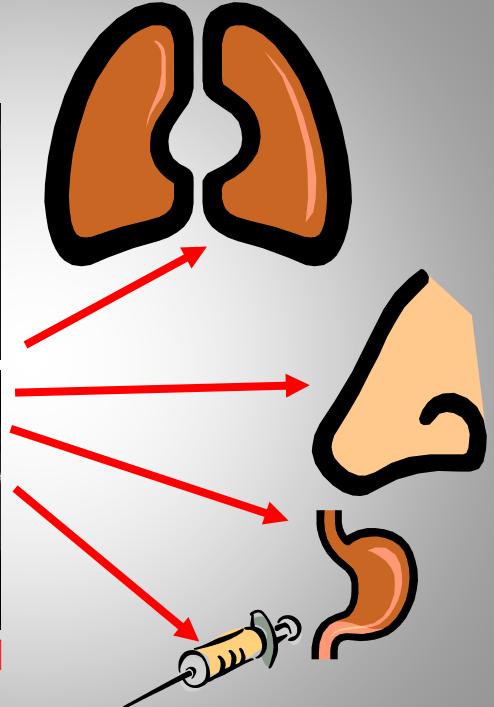


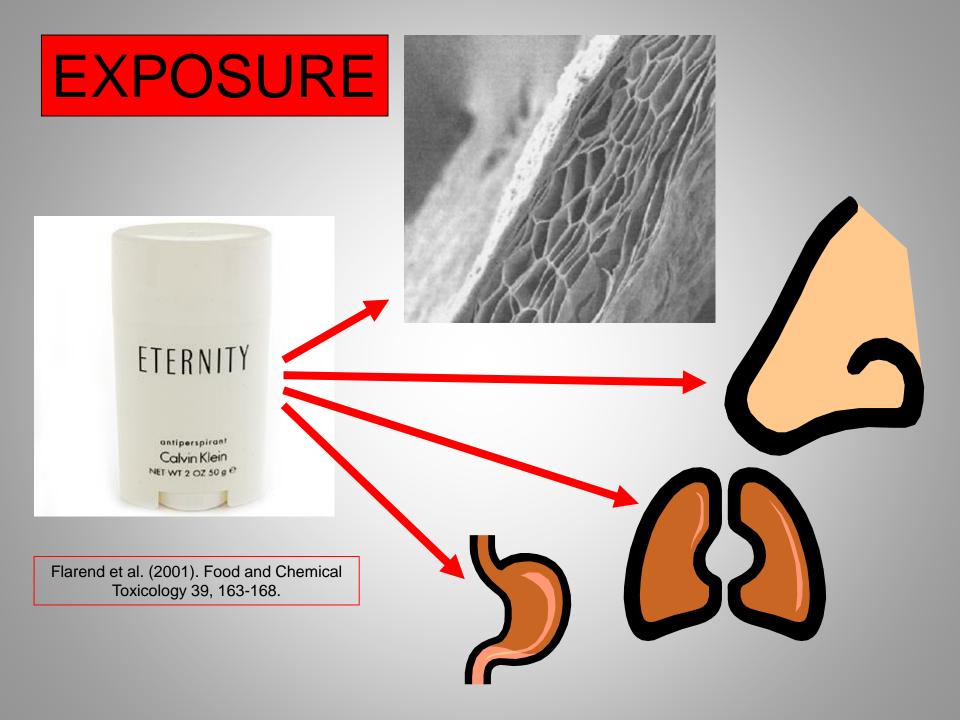
Exley et al. (2006). American Journal of Medicine 119, 276.e9-276.e11.

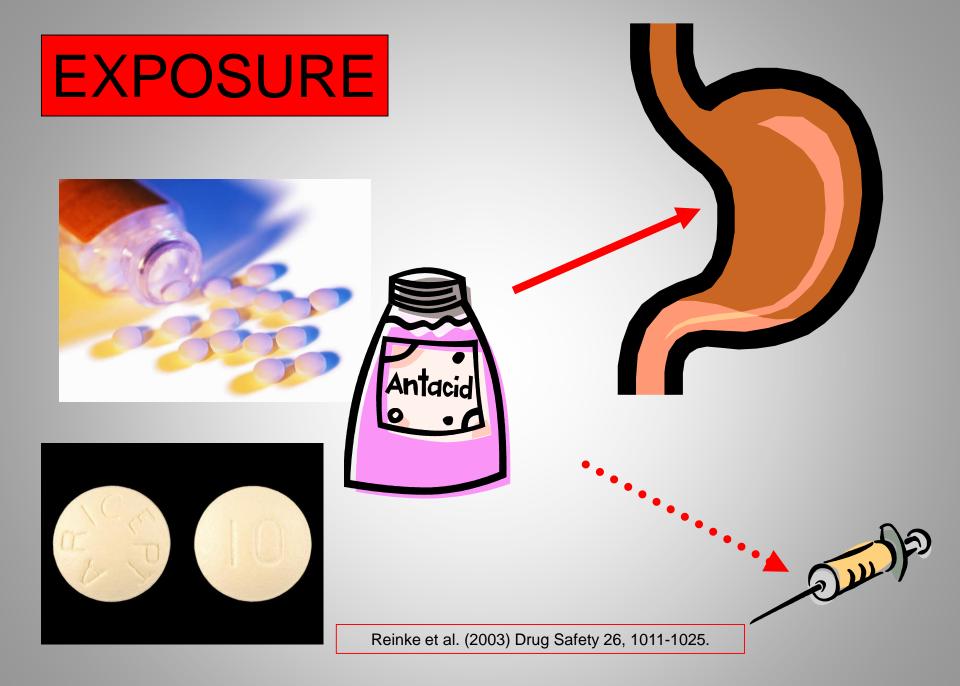




Exley et al. (2007) Addiction Biology 12, 197-199.

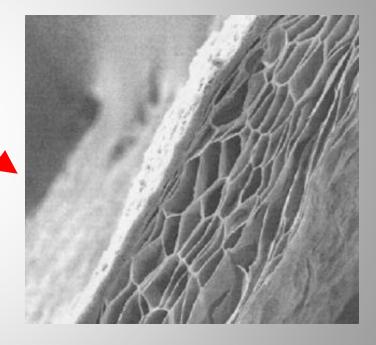


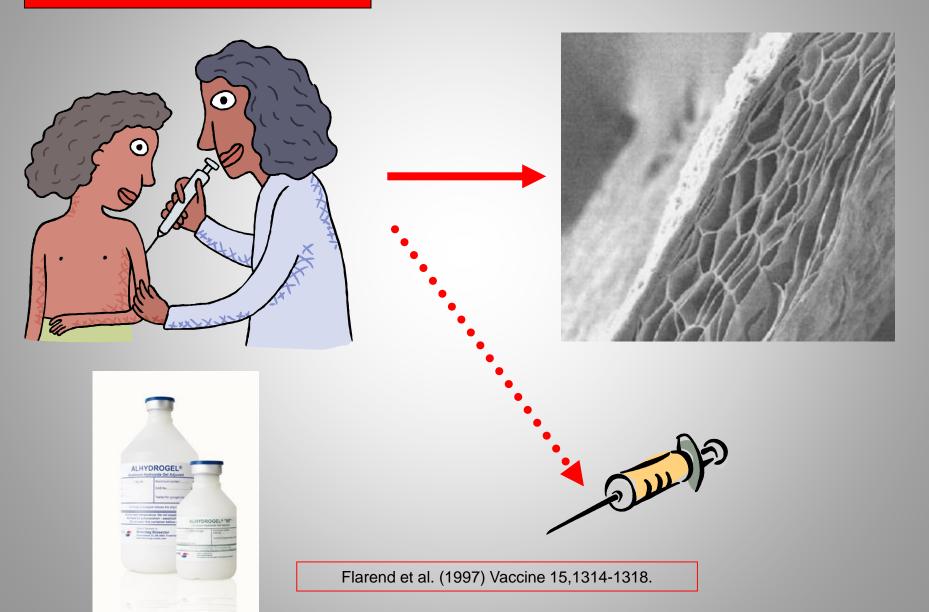




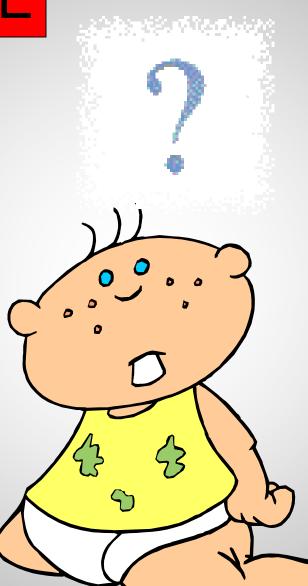


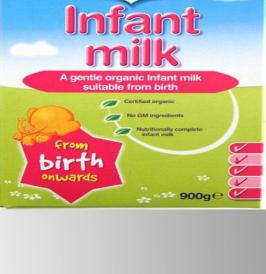
Nicholson & Exley (2007) Free Rad Biol Med 43, 1216-1217.











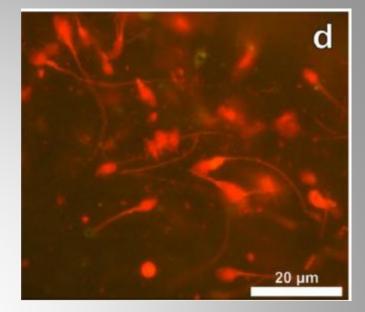
ORGANIC



Burrell & Exley (2010) BMC Pediatrics 10, 63. Chuchu et al. (2013) BMC Pediatrics 13, 162.



Human exposure to aluminium begins at conception!



Klein JP, Mold M, Mery L, Cottier M and Exley C (2014) Reproductive Toxicology 50, 43-48.

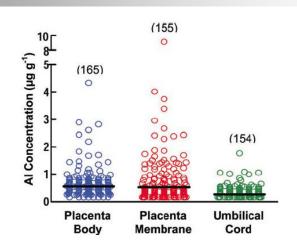
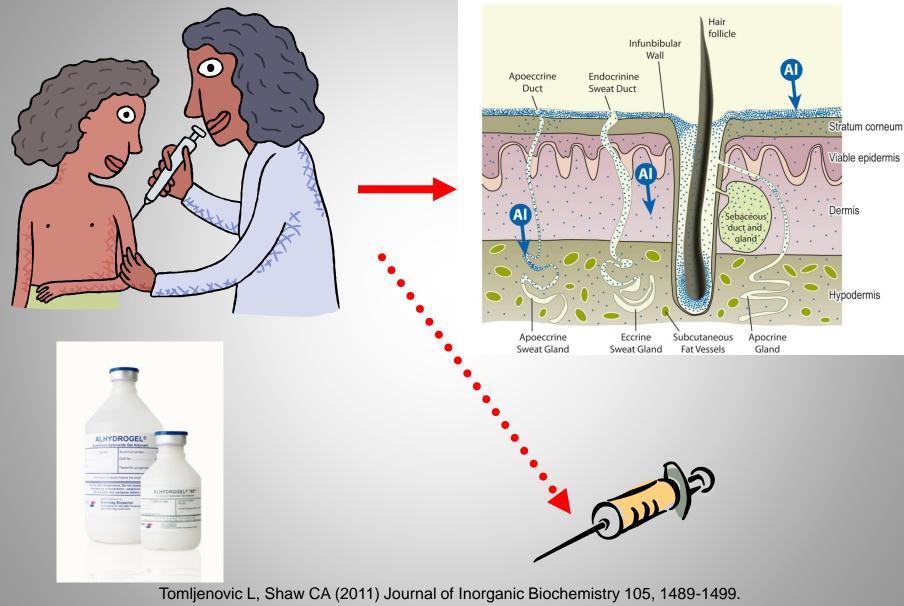


Fig. 3 Scatter dot plot showing geometric mean and range for Al $(\mu g g^{-1})$ in each placental tissue component. Each data point represents the average Al concentration measured by duplicate analysis of a sample. Black horizontal lines indicate geometric mean concentrations for each sample component. Numbers in parentheses indicate the number of placenta samples analyzed for each component.

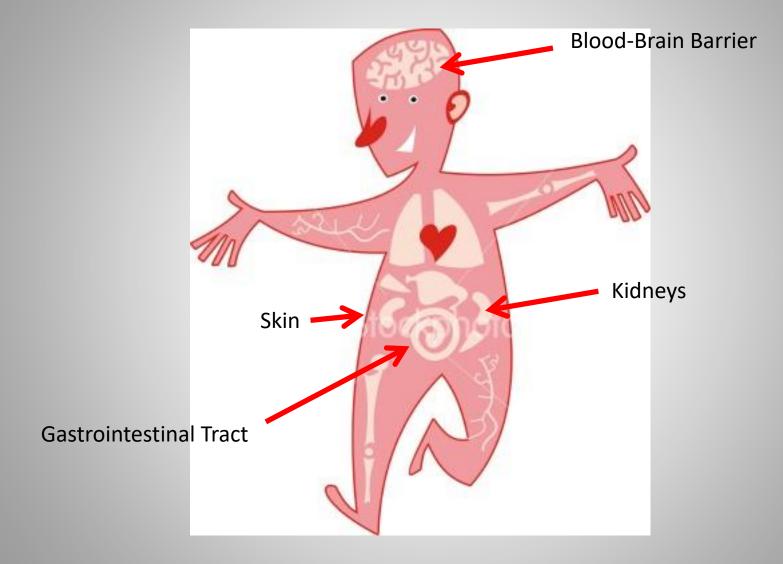
Kruger PC, Schell LM, Stark AD and Parsons PJ (2010) Metallomics 2, 621-627.

The Burden of Infant Vaccination



Tomljenovic L, Shaw CA (2012) Lupus 21, 223-230.

The Infant (and Especially the Neonate) is a Special Case





Dental Cements?

Demirkaya et al. (2016) Eur J Oral Sci 124, 75-81.* Nicholson JW & Czamecka B (2009) J Biomat Appl 24, 293-308



*..Al might have been released into the circulation from the three dental cements tested, especially MTA Angelus and MTA Fillapex.

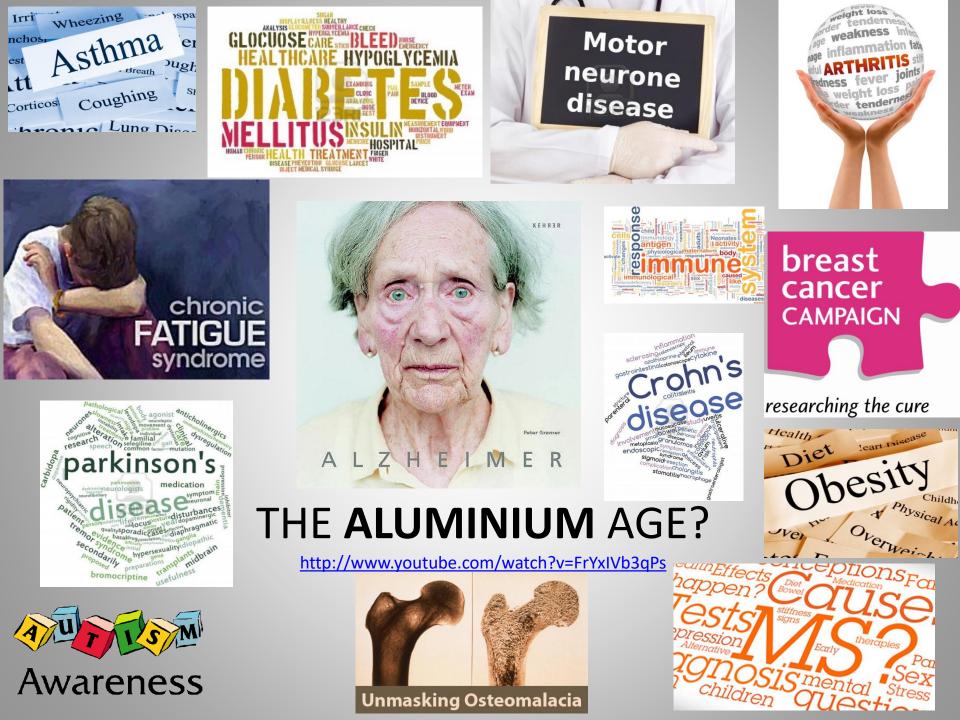
Toothpastes? Heidmann J & Polusen S (1997) Caries Res 31, 85-90. Rajwanshi et al. (1997) Sci Tot Environ 193, 243-249.

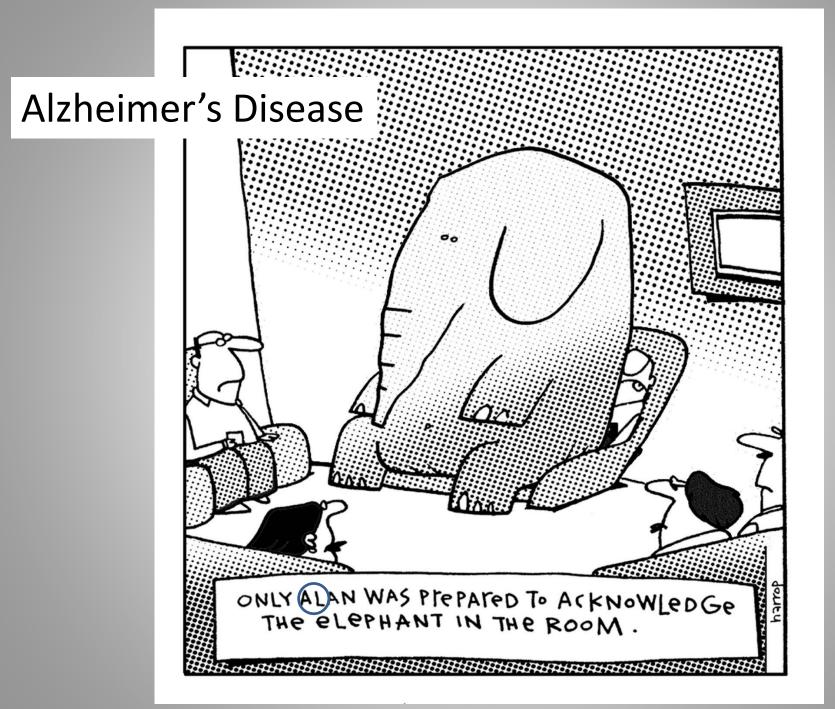
Living in 'The Aluminium Age' ensures our body burden of Al



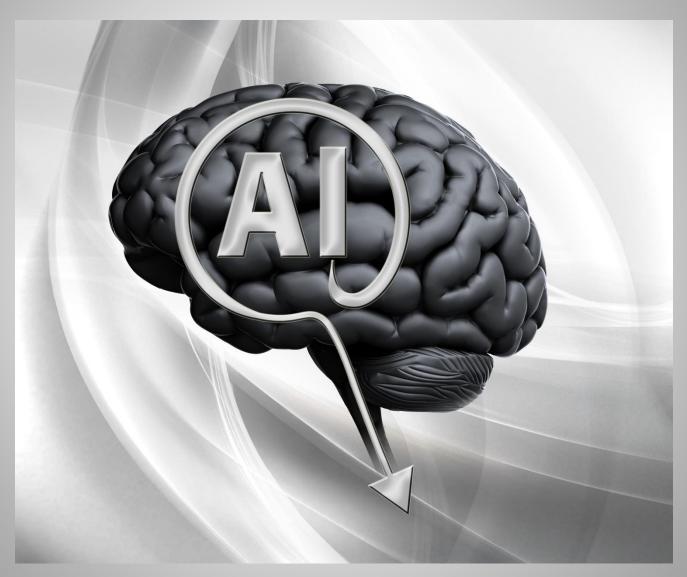


Minshall C et al. (2014) Journal of Trace Elements in Medicine and Biology 28, 87-88

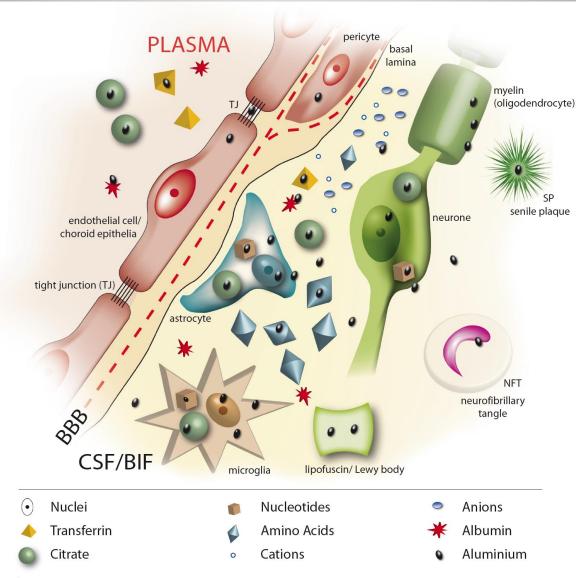




Aluminium and The Brain



Aluminium in the Brain



Exley C and House E (2011) Aluminium in the human brain. Monatshefte für Chemie - Chemical Monthly 142, 357-363.

Metallomics

Dynamic Article Links 🕟

Cite this: *Metallomics*, 2012, **4**, 56–65

www.rsc.org/metallomics

PAPER

Aluminium, iron and copper in human brain tissues donated to the medical research council's cognitive function and ageing study

Emily House,^{*a*} Margaret Esiri,^{*b*} Gill Forster,^{*c*} Paul G Ince^{*c*} and Christopher Exley^{**a*}

Sixty Human Brains >700 tissue digests (250 mg wet weight) 174 Method Blanks; Al = 54 ng Al/vessel (Mean + 1.654SD)

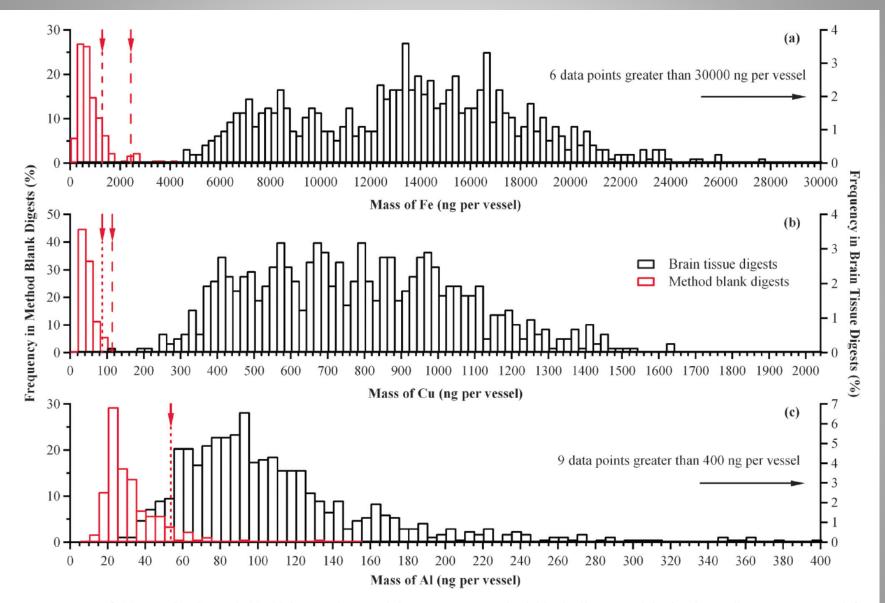


Fig. 2 Mass of (a) Fe, (b) Cu and (c) Al in vessels containing *ca* 5 mL method blank digests and brain tissue digests uncorrected for contamination. Arrows and dashed lines show the contaminant level which was subtracted from tissue digest values.

Aluminium Content of Human Brain

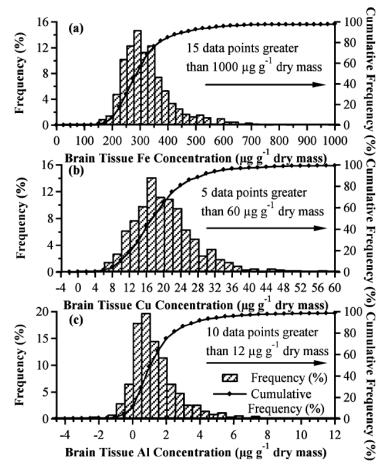


Fig. 3 Percentage frequency (bars) and cumulative frequency (line and marker) distributions of (a) Fe, (b) Cu and (c) Al concentrations in (n = 719, 720 and 713 respectively) brain tissues after subtraction of contamination.

The median Al content of tissues from all <u>60</u> <u>brains (n=713) is 1 μ g/g dry wt.</u>

In 52 out of 60 individuals at least one tissue sample exceeded 2 μ g Al/g dry wt.

In 41 out of 60 individuals at least one tissue sample exceeded 3.5 μ g Al/g dry wt.

Approximately 70% of individuals aged 70 – 103 years had at least one tissue Al content which should be considered as pathological.

House E, Esiri M, Forster G, Ince PG and Exley C (2012) Aluminium, iron and copper in human brain tissues donated to the medical research council's cognitive function and ageing study. Metallomics 4, 56-65.

Accidental Exposure to Aluminium

Camelford, Cornwall, United Kingdom, 1988

SHORT REPORT

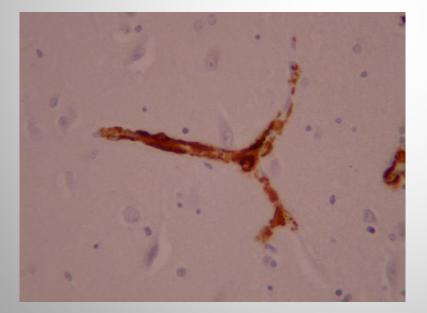
Severe cerebral congophilic angiopathy coincident with increased brain aluminium in a resident of Camelford, Cornwall, UK

C Exley, M M Esiri



See Editorial Commentary, p 811

J Neurol Neurosurg Psychiatry 2006;77:877-879. doi: 10.1136/jnnp.2005.086553



Frontal Cortex, n=5.

- 1. $23.00* \,\mu g/g \,dry \,wt.$
- 2. 3.24
- 3. 11.01
- 4. 4.33
- 5. 5.71

*Noted during measurement as a heavily vascularised piece of tissue.

Occupational Exposure to Aluminium

Exley and Vickers *Journal of Medical Case Reports* 2014, 8:41 http://www.jmedicalcasereports.com/content/8/1/41



CASE REPORT

Open Access

Elevated brain aluminium and early onset Alzheimer's disease in an individual occupationally exposed to aluminium: a case report

Mean Al content of frontal lobe tissue (n=46) is $2.98 (2.73) \mu gAl/g dry wt.$

Range is 0.00 (less than the method blank) to 12.97 μ gAl/g dry wt.

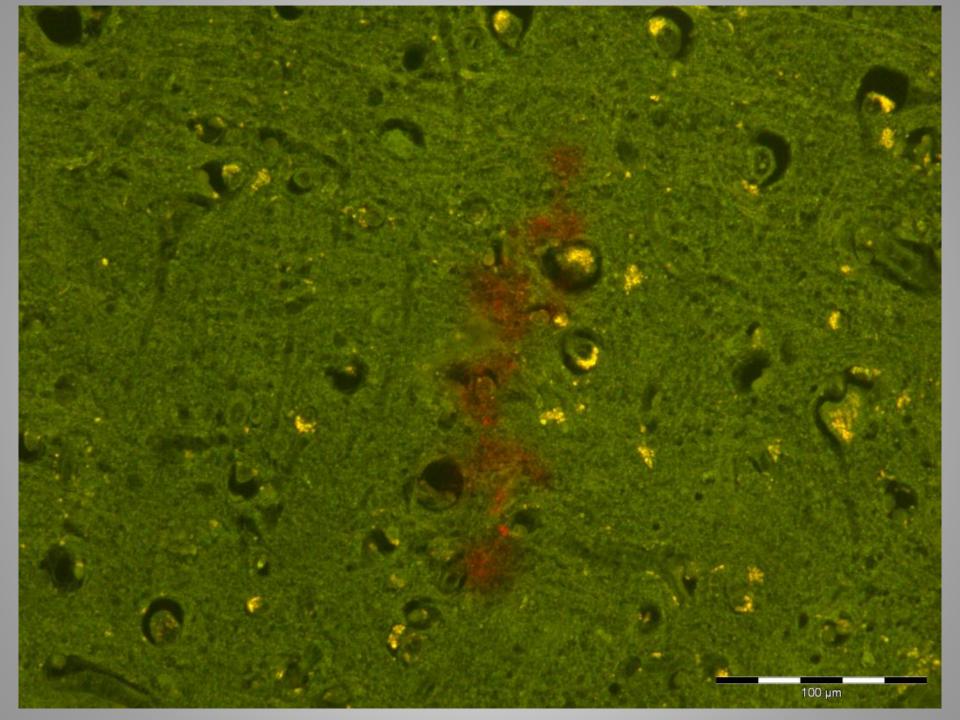
More than 30% of tissue samples had an Al content considered as pathological, greater than $3.50 \mu gAl/g dry wt$.

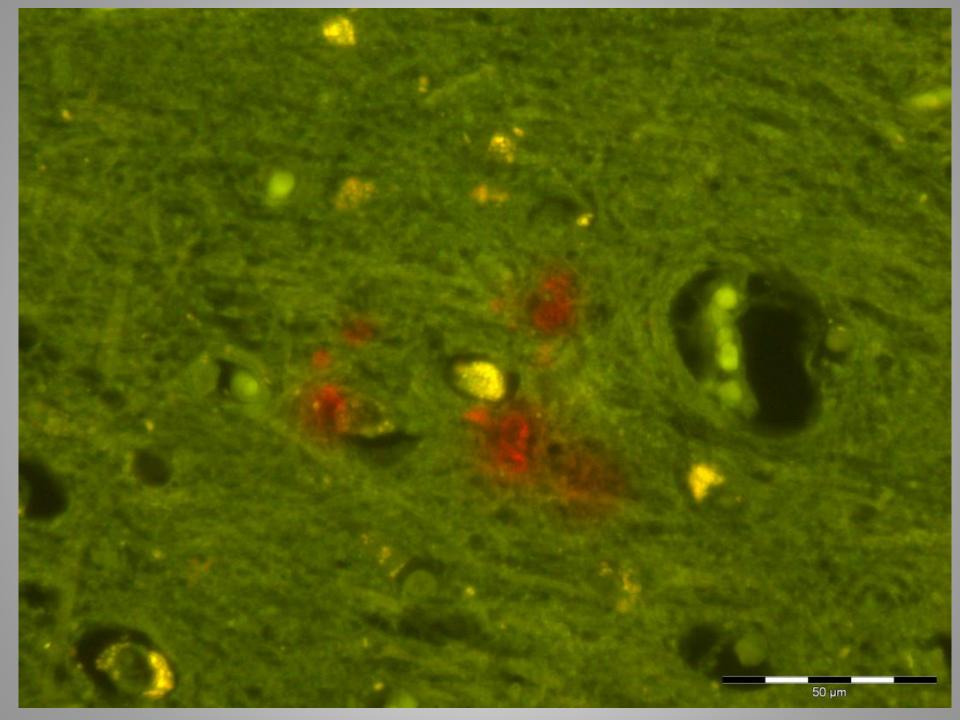
What about; Familial Alzheimer's Disease?

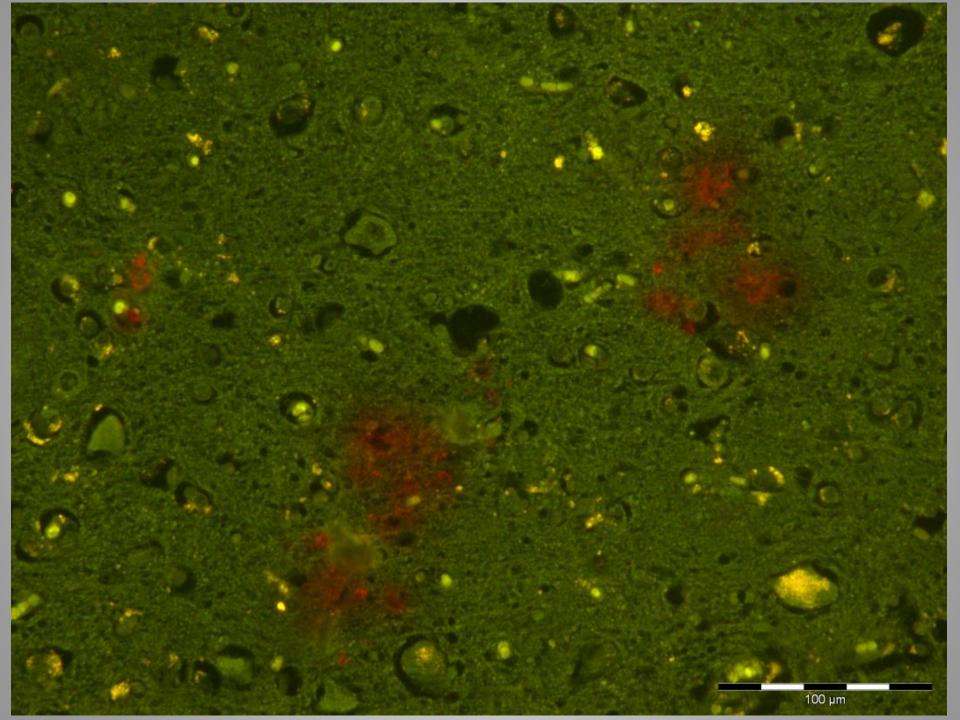
Summary indicating the lobe with the highest content of Al (Mean (SD) n=3)

(Pathologically Concerning/Pathologically Significant / 12 tissues)

Donor ID	Lobe	[Al] μg/g dry wt.
A1 0/0/12	Frontal	0.44(0.31)
A2 0/4/12	Occipital	9.99(1.61)
A3 6/3/12	Occipital	4.29(3.21)
A4 3/2/12	Parietal	3.30(4.43)
A5 1/2/12	Frontal	7.05(6.10)
A6 1/3/12	Occipital	9.57(14.08)
A7 2/0/12	Temporal	1.81(0.78)
A8 3/6/12	Frontal	14.41(18.44)
A9 4/2/12	Frontal	3.81(4.11)
A10 5/1 /12	Occipital	2.97(1.01)
A11 2/2/12	Occipital	9.31(12.71)
A12 2/3 /12	Temporal	2.91(2.89)







Complacency? Aluminium in Drugs for Alzheimer's Disease



The Al content of Reminyl (Galantamine hydrobromide) is approximately 600 µg/g. When a single tablet is added to 50 mL of a simulated stomach solution (0.25% *w/v* sodium lauryl sulphate, 0.05% w/v sodium azide, 35mM sodium chloride, 5mL 15.8M HNO₃ and ultapure water, pH 1.5 – 1.7) and incubated for 1h at 37°C the tablet dissolves giving an orange solution*. The total [Al] of this solution is <u>1300.0 (76.7) µg/L (n=9).</u>

*Contains orange-yellow S-aluminium lake, E110

Multiple Sclerosis

ARTICLE

Multiple Sclerosis 2006; 12: 533-540

Elevated urinary excretion of aluminium and iron in multiple sclerosis

Christopher Exley¹, Godwin Mamutse², Olga Korchazhkina³, Eleanor Pye², Stanislav Strekopytov¹, Anthony Polwart⁴ and Clive Hawkins²

Study title:	Urinary excretion of aluminium and silicon in secondary progressive multiple sclerosis.
REC reference:	14/YH/1115
Amendment number:	Substantial Amendment 1
Amendment date:	28 November 2014
IRAS project ID:	144340

Study title:	Urinary excretion of aluminium and silicon in secondary progressive multiple sclerosis.
REC reference:	14/YH/1115
Amendment number:	Substantial Amendment 1
Amendment date:	28 November 2014
IRAS project ID:	144340

Primary objective – measurement of urinary excretion of aluminium, iron and silicon over two consecutive13 week periods to determine if the body burden of aluminium is influenced by drinking a silicon-rich mineral water.

Secondary objective – measurements of physical, cognitive and quality of life parameters in order to identify any clinical changes in SPMS patients.

Clinical trial complete and data are being processed, RESULTS ARE IMMINENT!



Burrell S-A M and Exley C (2010) BMC Pediatrics 10:63 Chuchu N, Patel B, Sebastian B and Exley C (2013) BMC Pediatrics 13, 162.

Study title:	The contribution of breast and formula feeding to the body burden of aluminium in infants (0-12 months)
REC reference:	14/WM/1114
Amendment number:	1
Amendment date:	09 December 2015
IRAS project ID:	148651

Primary Objective: The research will monitor the urinary excretion of aluminium in term infants fed breast milk, infant formula and term infants fed combinations of the two over the first 12 months of their lives. Additional variables will include urine creatinine (Crt) and urinary excretion of silicon (Si).

The research will additionally monitor aluminium levels through the collection of hair samples as a secondary indicator. Both methods will be used for measurement of the body burden of aluminium in infants. The data collected will be then used to establish if there are any relationships between the body burden of aluminium and the type of infant feeding.

Aluminium Adjuvants in Vaccines and Immunotherapy

How Do They Really Work?!

There are no <u>clinically-approved</u> (aluminium) adjuvants!

There are only clinically-approved vaccines.

The safety of adjuvants is established alongside the safety of vaccines.

So, why are aluminium adjuvants used as placebos in vaccine trials?!

So, what do we know about aluminium adjuvants which are used in clinicallyapproved vaccines?

Alhydrogel

Aluminium oxyhydroxide (boehmite)

Poorly crystalline – hydrated structure (14.1% H_2O at the surface interface)

Composed of nanoneedles - 4.5 nm \times 2.2 nm \times 10 nm in size

Most frequently used adjuvant in commercial vaccines



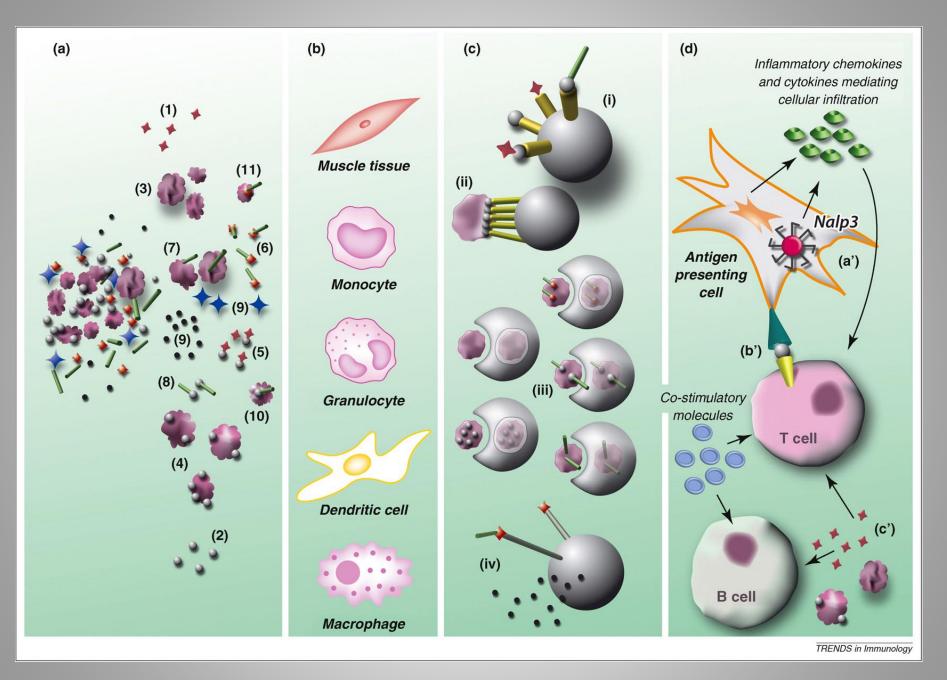
Aluminium hydroxyphosphate

Amorphous – hydrated structure (24.2% H_2O at the surface interface)

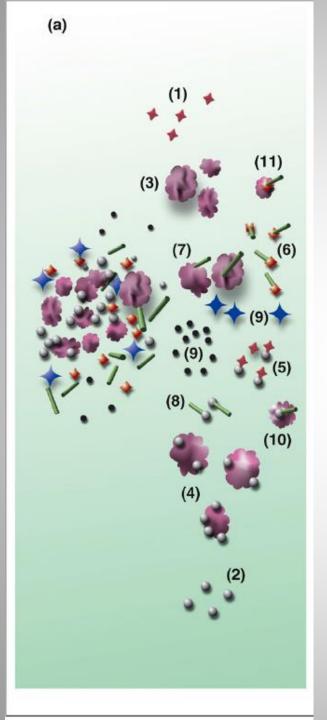
Composed of platy particles – 50nm in size

So, how do aluminium adjuvants work?

How might understanding this also begin to explain the known adverse events associated with their use in vaccines?



Exley et al., (2010) The Immunobiology of aluminium adjuvants; how do they really work? Trends in Immunology, 31,103-109



The Critical Environment of the Injection Site

What Happens to the Aluminium Adjuvant?

Conclusions

In 0.9% NaCl, negatively charged Adju-Phos has a larger overall particle size than positively charged Alhydrogel

♦ Alh - ~
$$72\%$$
 ≤ 2.7µm

* Adj - ~ $28\% \leq 2.7 \mu m$

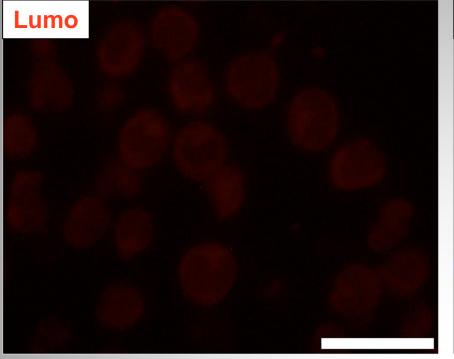
➤ At the site of injection both adjuvants become negatively charged upon administration

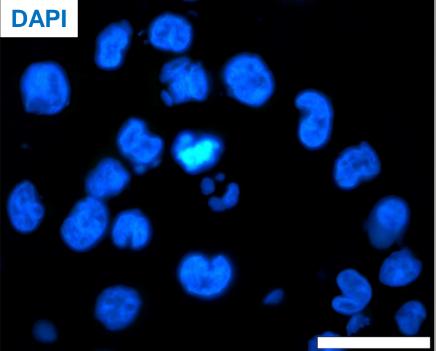
➢ Following administration Alhydrogel has a larger abundance of particles available for phagocytosis.

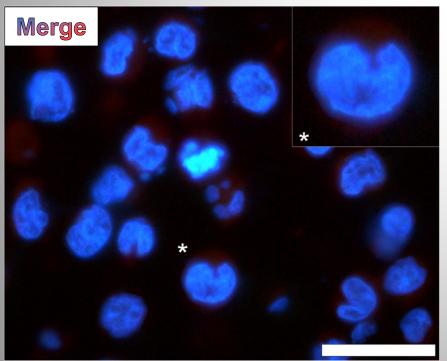
• Alh - ~
$$97\% \leq 2.7 \mu m$$

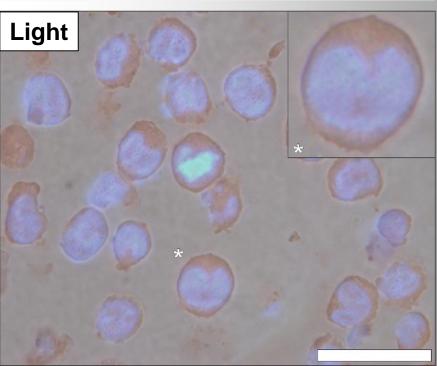
What About the Cellular Response to Aluminium Adjuvants?

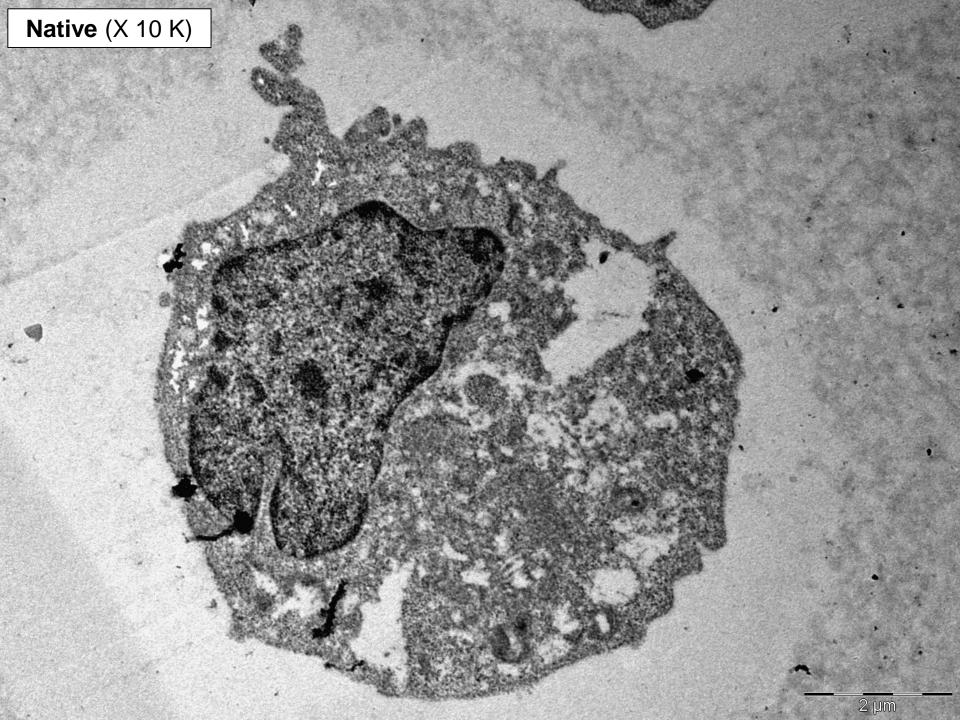
Native THP-1 cells (R10)











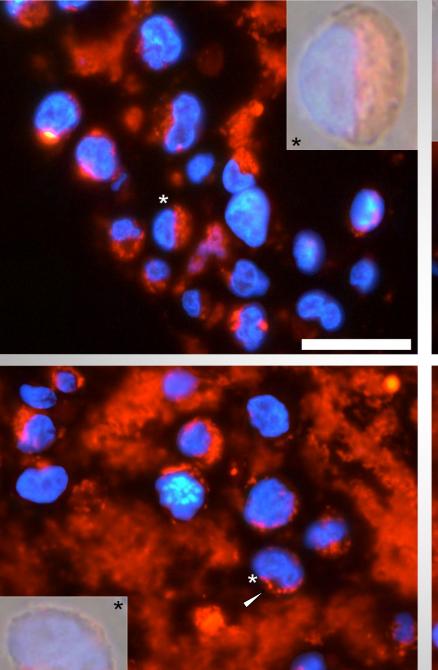
Alhydrogel®

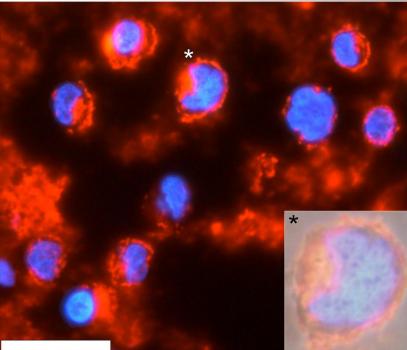
2.5 - 100 μg/mL

50µg/mL

/

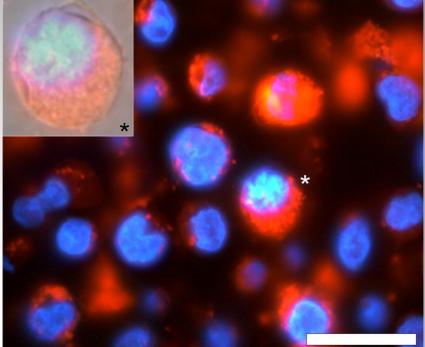








100µg/mL



50µg/mL Alh (X 8 K)

7

5 µm

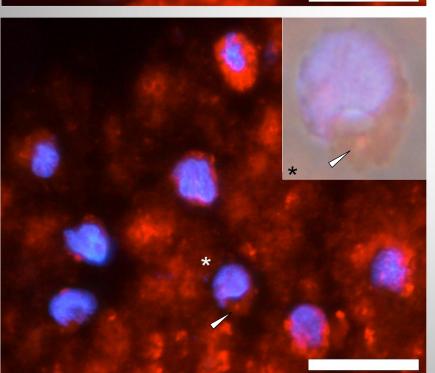
50µg/mL Alh (X 30 K)

1 µm

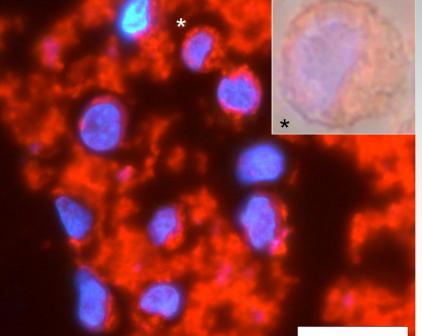
Adju-Phos[®]

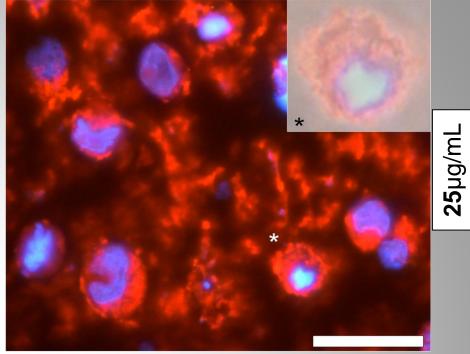
2.5 - 100 μg/mL

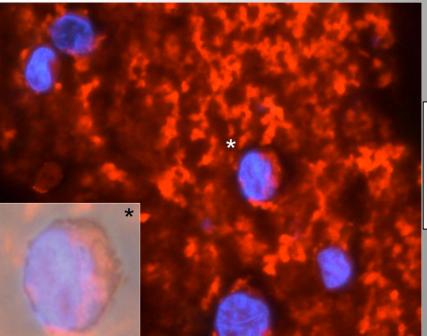




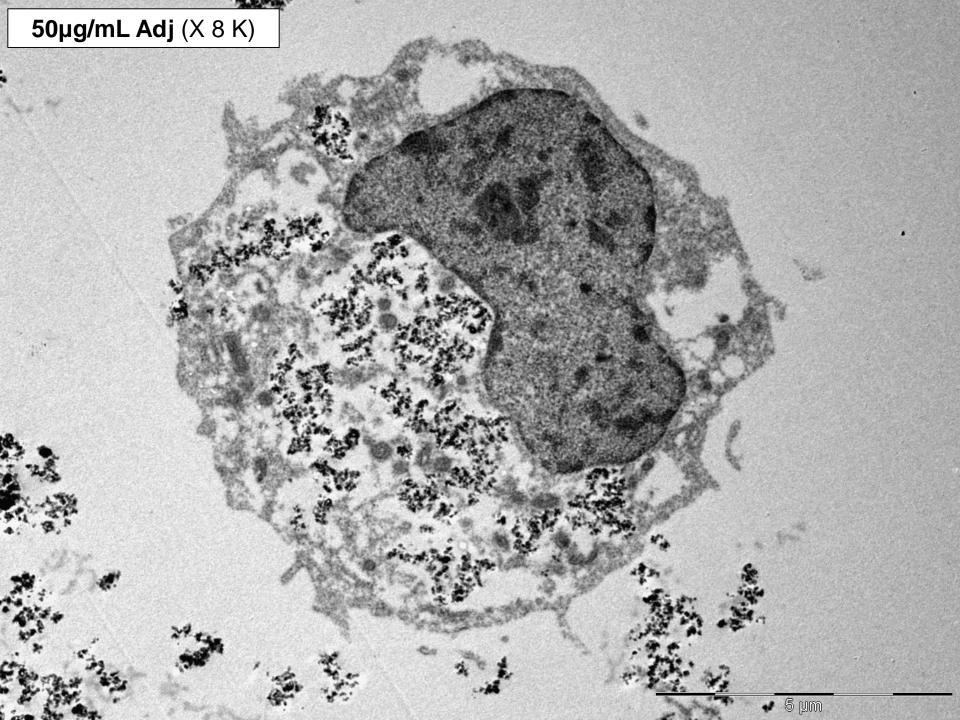








100µg/mL



50µg/mL Adj (X 30 K)

1 µm

So, What About the Toxicity of Aluminium Adjuvants?

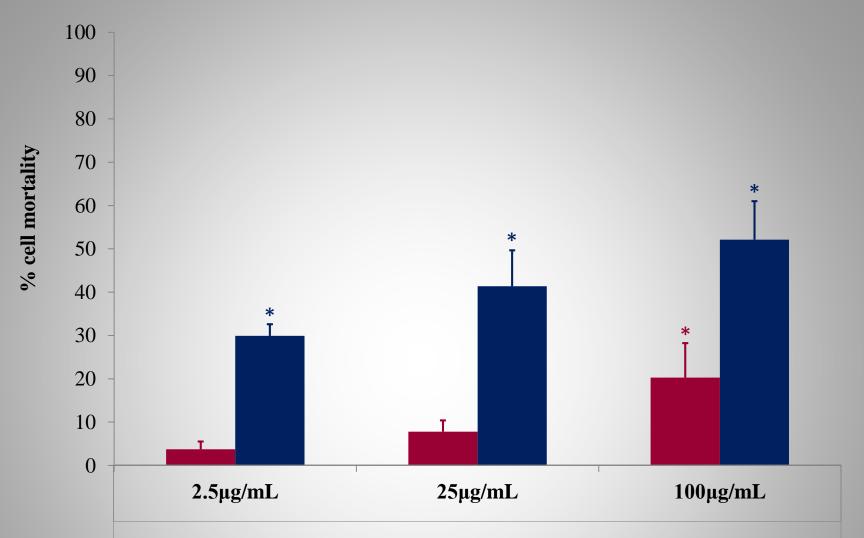


Fig 9: The % mortality experienced in THP-1 cell populations upon exposure to various concentrations of aluminium adjuvants relative to the control group, as elucidated using the live/dead cytotoxicity assay. Plum and blue bars represent Alhydrogel and Adju-Phos respectively. Error bars are representative of ±SD of 3 individual replicates and statistical significance is shown between treatments and respective control groups

Conclusions

- For the two aluminium adjuvants used in clinically approved vaccines, intracellular particulates of Alhydrogel[®] and Adju-Phos[®], were observed localised in cell cytoplasm only.
- Only co-culture with Adju-Phos[®] resulted in the release of extracellular genetic material.
- The experimental Imject[™] Alum adjuvant also exhibited particulates localised in the cell cytoplasm of THP-1 cells, typically larger in size.

Conclusions cont.

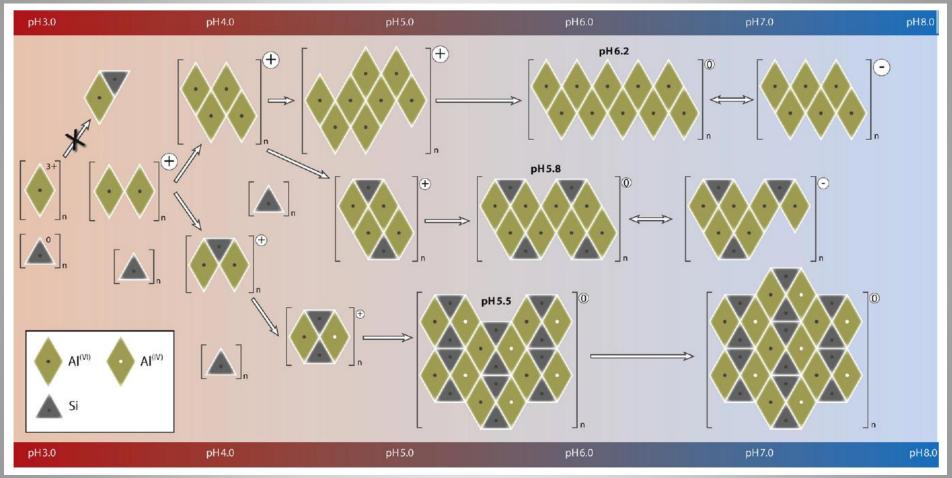
- Higher concentrations of aluminium adjuvants cocultured with THP-1 cells were observed to result in their reduced cellular uptake (50 & 100µg/mL Adju-Phos[®] and 100µg/mL Imject[™] Alum).
- The cytotoxicities of the two aluminium adjuvants used in clinically-approved vaccines are significantly different with Adju-Phos® expected to induce greater toxicity at the injection site.
- The observed lower toxicity of Alhydrogel[®] despite its high intracellular burden may predispose this adjuvant to its translocation to sites away from the injection site.

Human Exposure to Aluminium

An Effective Therapeutic Strategy?



The Unique Inorganic Chemistry of the Reaction of Aluminium with Silicic acid



Coordination Chemistry Reviews 256 (2012) 82-88

A Bioinorganic Solution to Aluminium-Related Disease?

1989

Acute toxicity of aluminium to fish eliminated in silicon-rich acid waters

J. D. BIRCHALL, C. EXLEY, J.S. CHAPPELL & M. J. PHILLIPS

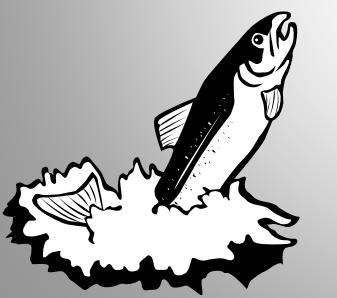
Nature 338, 146 - 148 (09 March 1989); doi:10.1038/338146a0

2006

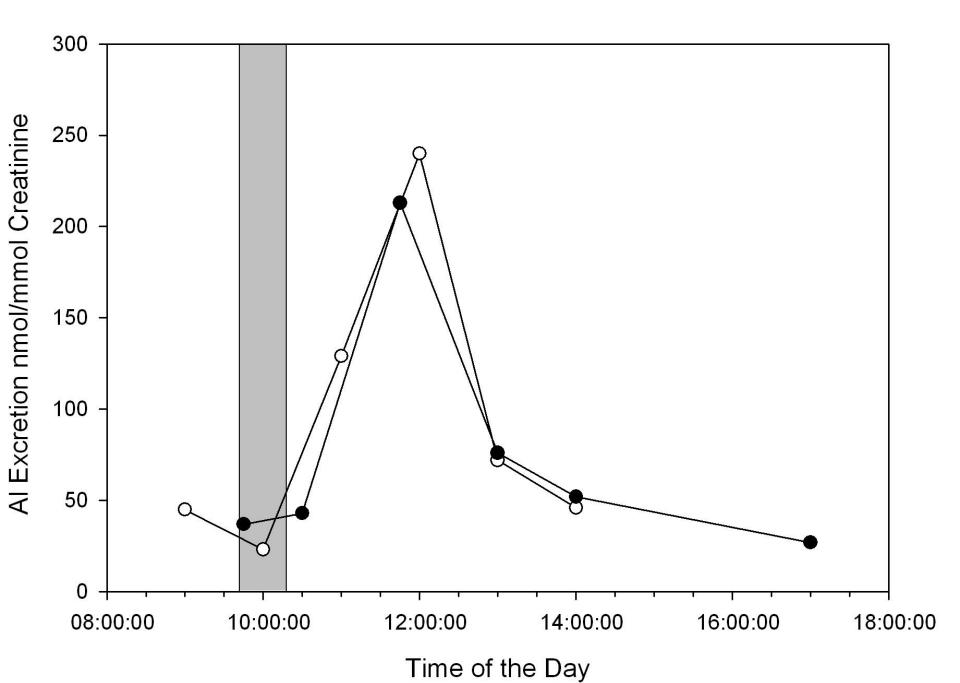
Non-invasive therapy to reduce the body burden of aluminium in Alzheimer's disease

Christopher Exley, Olga Korchazhkina, Deborah Job, Stanislav Strekopytov, Anthony Polwart and Peter Crome

Journal of Alzheimer's Disease 10 (2006) 17–24









Ten individuals with Alzheimer's disease were asked to drink up to 1.5L of Volvic mineral water each day on five consecutive days as part of their everyday diets. The individuals were asked to collect their first urine sample of each day for both the five days preceding the trial (not drinking Volvic) and the five days of the trial.

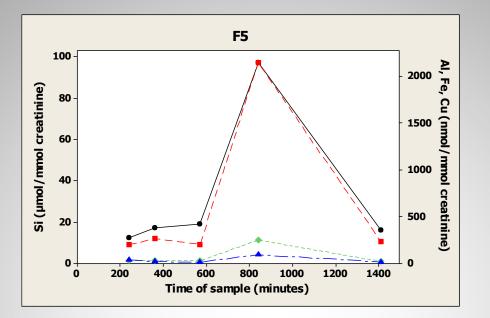
Urine samples, pre and post Volvic, were analysed for AI, Fe and Si.

CONCLUSION

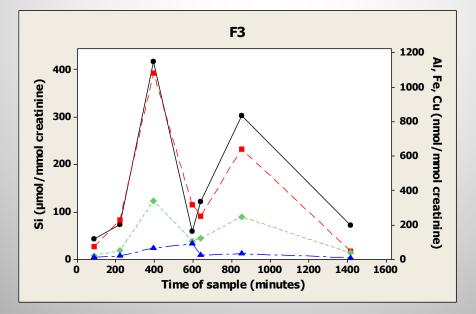
In individuals with Alzheimer's disease;

Drinking Volvic mineral water increased significantly (P<0.001) urinary excretion of silicic acid (34.3 ± 15.2 to $55.7 \pm 14.2 \mu$ mol/mmol creatinine) and concomitantly reduced significantly (P=0.037) urinary excretion of aluminium (86.0 ± 24.3 to 62.2 ± 23.2 nmol/mmol creatinine). The latter was achieved without any significant (P>0.05) influence upon the urinary excretion of iron (20.7 ± 9.5 to 21.7 ± 13.8 nmol/mmol creatinine).

The reduction in urinary aluminium supported the future longer-term use of silicic acid as non-invasive therapy for reducing the body burden of aluminium in Alzheimer's disease.







Silicon-Rich Mineral Water as a Non-Invasive Test of the 'Aluminum Hypothesis' in Alzheimer's Disease

Samantha Davenward^a, Peter Bentham^b, Jan Wright^b, Peter Crome^c, Deborah Job^c, Anthony Polwart^d and Christopher Exley^{a,*}

^aThe Birchall Centre, Lennard-Jones Laboratories, Keele University, Stoke-on-Trent, Staffordshire, UK ^bBirmingham and Solihull Mental Health NHS Foundation Trust, The Barberry Centre, Birmingham, UK ^cNorth Staffordshire Combined Healthcare NHS Trust, Harplands Hospital, Stoke-on-Trent, UK ^dLife Sciences, Keele University, Stoke-on-Trent, Staffordshire, UK

We have provided preliminary evidence that over 12 weeks of silicon-rich mineral water therapy the body burden of aluminium fell in individuals with Alzheimer's disease and, <u>concomitantly, cognitive performance</u> <u>showed clinically relevant improvements in at least 3 out of 15</u> <u>individuals.</u>

The Twelfth Keele Meeting on Aluminium (Keele12)

4-8th March 2017 Pinnacle Hotel, Vancouver, Canada